

41 DETERMINATION OF LEAD CONTENT IN ALCOHOLIC BEVERAGES BY ATOMIC ABSORPTION SPECTROMETRY	Page 1 of 3
<div> Division of Forensic Science </div> <div> TOXICOLOGY TECHNICAL PROCEDURES MANUAL </div>	Amendment Designator:
	Effective Date: 31-March-2004
<div> 41 DETERMINATION OF LEAD CONTENT IN ALCOHOLIC BEVERAGES BY ATOMIC ABSORPTION SPECTROMETRY </div> <div> 41.1 Summary </div> <div> 41.1.1 The determination of lead content by flame atomic absorption spectrometry is used to help identify illegally distilled spirits. Samples are heated to high temperatures (approx 2300° C) in an acetylene/air flame. The flame dissociates chemical bonds and releases free metal atoms which absorb light. The amount of absorbed radiant energy is a function of concentration of lead present in the sample. </div> <div> 41.2 Reagents </div> <div> 41.2.1 95% ethanol (grain alcohol) </div> <div> 41.2.2 Lead oxide, 1000 ppm in 0.3 M nitric acid </div> <div> 41.2.3 Illegally distilled spirit control </div> <div> 41.2.4 Legally distilled spirit control (vodka) </div> <div> 41.3 Solutions and standards </div> <div> 41.3.1 40% ethanol. Pipet 421 mL 95% ethanol into a 1 L volumetric flask and QS to volume with dH₂O. </div> <div> 41.3.2 5.0 ppm lead standard. Pipet 5 mL of 1000 ppm lead standard into a 1 L volumetric flask and QS to volume with 40% ethanol. </div> <div> 41.3.3 0.5 ppm lead standard. Pipet 10 mL of the 5.0 ppm lead standard into a 100 mL volumetric flask and QS to volume with 40% ethanol. </div> <div> 41.3.4 1.0 ppm lead standard. Pipet 20 mL of the 5.0 ppm lead standard into a 100 mL volumetric flask and QS to volume with 40% ethanol. </div> <div> 41.3.5 2.0 ppm lead standard. Pipet 40 mL of the 5.0 ppm lead standard into a 100 mL volumetric flask and QS to volume with 40% ethanol. </div> <div> 41.4 Apparatus </div> <div> 41.4.1 Pipets, Class A, 5 mL, 10 mL and 20 mL volume </div> <div> 41.4.2 Flame Atomic Absorption Spectrometer (Varian SPECTRAA 55B) </div> <div> 41.4.2.1 Parameters. Instrument parameters may be adjusted to improve performance. </div> <div> <div> 41.4.2.1.1 Instrument mode: </div> <div> Absorbance Emission </div> </div> <div> <div> 41.4.2.1.2 Active lamp: </div> <div> 2 </div> </div> <div> <div> 41.4.2.1.3 Active current: </div> <div> 8 </div> </div> <div> <div> 41.4.2.1.4 Standby current: </div> <div> 0 </div> </div>	

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<div> <div>41.4.2.1.5</div> <div>D2 correction:</div> <div>Yes</div> </div> <div> <div>41.4.2.1.6</div> <div>Gas type:</div> <div>Air acetylene</div> </div> <div> <div>41.4.2.1.7</div> <div>Wavelength:</div> <div>Other</div> </div> <div> <div>41.4.2.1.8</div> <div>Other:</div> <div>217 nm</div> </div> <div> <div>41.4.2.1.9</div> <div>Slit:</div> <div>1.0</div> </div>		
41.5 Procedure <div> <div>41.5.1</div> <div>Ensure AA spectrophotometer exhaust system is on.</div> </div> <div> <div>41.5.2</div> <div>Turn on gas supplies in the following order:</div> <div> <div>41.5.2.1</div> <div>Air at 50-60 psi</div> </div> <div> <div>41.5.2.2</div> <div>Acetylene at 12-15 psi</div> </div> </div> <div> <div>41.5.3</div> <div>Turn on AA power. Load method (1 Pb 217).</div> </div> <div> <div>41.5.4</div> <div>Ignite flame.</div> </div> <div> <div>41.5.5</div> <div>Allow spectrophotometer to warm up for at least 15 minutes.</div> </div> <div> <div>41.5.6</div> <div>Aspirate lead free 40% ethanol (blank) and zero instrument.</div> </div> <div> <div>41.5.7</div> <div>Aspirate each lead calibrator solution. Read results, calibrate instrument and record absorbance on lead worksheet.</div> </div> <div> <div>41.5.8</div> <div>Aspirate samples and read results. The AA will take 3 readings and average the three absorbances and calculate lead concentration (in ppm).</div> </div> <div> <div>41.5.9</div> <div>Record the absorbances and concentration (ppm) of each sample on the lead worksheet.</div> </div> <div> <div>41.5.10</div> <div>Repeat 41.5.8 and 41.5.8 on case samples.</div> </div> <div> <div>41.5.11</div> <div>After all samples have been run, rerun at least one calibrator as a sample and record on worksheet.</div> </div>		
41.6 Interpretation <div> <div>41.6.1</div> <div>The tolerable lead limit for legally produced beverages is 0.2 ppm. Levels equal to or greater than 0.3 ppm indicate the beverage was illegally distilled.</div> </div> <div> <div>41.6.2</div> <div>Lead contents in excess of 3.5 ppm can be hazardous to consumers.</div> </div>		
41.7 References <div> <div>41.7.1</div> <div>AOAC 14th edition, 25.115 (p 464), 1984</div> </div> <div> <div>41.7.2</div> <div>Varian Spectrometer AA 50/55 Instrument Manual</div> </div>		

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<p>41.7.3 The Pharmacological Basis of Therapeutics. Goodman and Gilman. 5th edition, p 939</p>	